

NUCLEOPHILIC DISPLACEMENT REACTIONS WITH THIOBENZOATE.  
NEW SYNTHESSES OF DEOXY-THIOADENOSINE DERIVATIVES STARTING FROM ADENOSINE.

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Modifications of the carbohydrate moiety of adenosine have led to compounds such as 3'-amino-3'-deoxyadenosine, cordycepin (3'-deoxyadenosine) and 4'-thioadenosine<sup>1)</sup>, which show cell growth inhibitory activity. These findings also prompted the interesting synthesis of 3'.deoxy-3'-thioadenosine (6), which was prepared previously by Goodman et al.<sup>2)</sup> as follows: methyl-3,5-di-O-tosyl- $\alpha$ -D-xylofuranoside was converted to the 2-thionobenzoate on reaction with thiobenzoylchloride. Some 2-benzoate was also formed. Neighbouring group participation resulted in intramolecular 3-sulfonate displacement, and further steps yielded 1-O-acetyl-3-S-benzoyl-3-thio- $\beta$ -D-ribofuranose. The latter was converted in 18 % yield to 6 by the chloromercury procedure.

We were interested in 6 as starting material for the preparation of the 3'-thio analog of c-AMP, and in the following report we describe a more convenient route to this compound.

Direct displacement reactions of the functionalized secondary hydroxyl groups of furanoses are generally considered to be difficult<sup>3)</sup>, but recent examples have been described<sup>4)</sup>. 3'-Deoxy-3'-iodo-xylofuranosyladenine (3) can be prepared from adenosineriboepoxide (2)<sup>5)</sup>, which is readily available from adenosine (1)<sup>6)</sup>. When 3 was reacted with sodium thiobenzoate in DMF at 100°C, two main products were formed. The major one (50 % yield) was readily identified as the 3',4'-unsaturated compound 5<sup>7)</sup>. The second compound (40 %) contained a S-benzoyl as well as an O-benzoyl group. The analytical and spectroscopic data are in agreement with structure 4. (Mass spectrum:  $M^+$  = 491 m/e, n.m.r. see table 1).

Under the employed reaction conditions O-benzoylation at the 2'-OH also takes place, as was indicated by the low field shift of C<sub>2'</sub>-H in the n.m.r.. Upon catalytic hydrogenation, compound 4 yielded 2'-O-benzoyl-3'-deoxyadenosine (7), which with sodium methoxide gave 3'-deoxyadenosine (8). With adenosine, xylo-adenosine and 3'-deoxyadenosine O-benzoylation did not occur with sodium thio-benzoate. Therefore we assume that after the nucleophilic displacement, 3'-S → 2'-O benzoyl migration has occurred, followed by benzoylation of the 3'-SH group<sup>8)</sup>. This explanation is supported by the reaction of 6 with thiobenzoate to yield 4. Compound 6 was obtained on treating 4 with sodium methoxide in the presence of nitrogen to avoid disulfide formation. The mass spectrum of 6 was identical with the one previously published<sup>9)</sup>. The physical data differed slightly from the reported<sup>2)</sup>: m.p. 204 - 206°C<sup>10)</sup>;  $[\alpha]_D^{25} = -18^\circ$  (c = 0,20 in H<sub>2</sub>O). The SH-determination with "Ellman's Reagent" [5,5'-dithiobis-(2-nitro-benzoic acid)]<sup>11)</sup> gave the correct value.

In the n.m.r. spectrum the coupling constant J<sub>1',-2'</sub> (2,0 Hz) of 6 was found to be similar to that of xyloadenosine (J<sub>1',-2'</sub> = 2,0 Hz)<sup>6)</sup>.

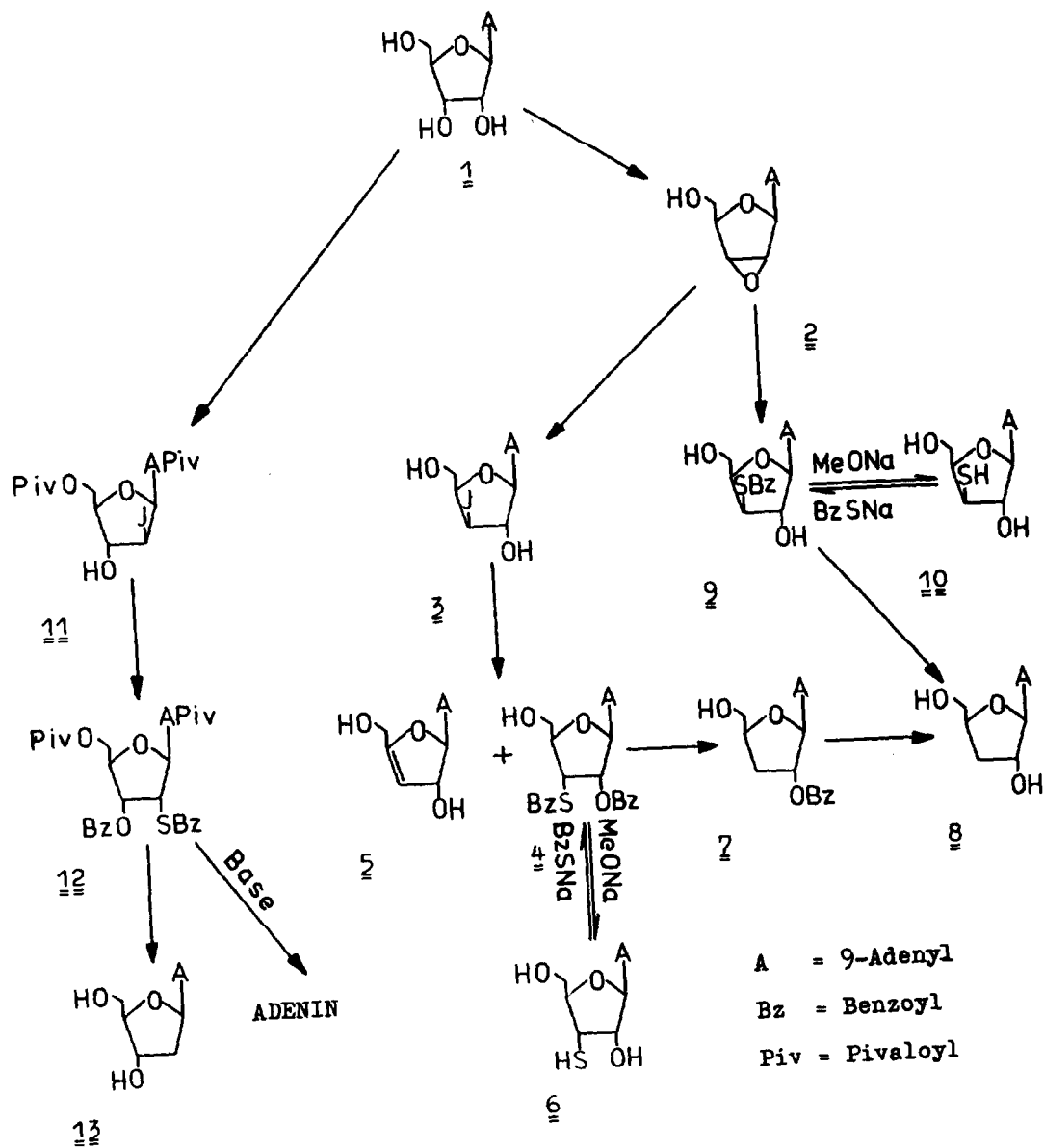
For comparison the corresponding xylo-thioadenosine (10) was prepared. Adenosineriboepoxide (2) was selectively ring opened with thiobenzoic acid, yielding 9. Catalytic hydrogenation of 9 with Pd/C gave 3'-deoxyadenosine (8), which proves the ring opening has taken place at the 3'-position. Sodium methoxide treatment of 9 under nitrogen gave 10. The J<sub>1',-2'</sub> coupling constant of the xylo-derivative 10 was similar to that of adenosine (J<sub>1',-2'</sub> = 6,1 Hz). When 10 was reacted with sodium thiobenzoate, the SH-group was selectively benzoylated yielding 9. This is consistent with trans arrangement of the 3'-SH and 2'-OH group, which excludes 3'-S → 2'-O-benzoyl migration.

All these findings suggest that the configuration of 6 is ribo and that of 10 is xylo. The 2',3'-endo:exo ratio of the ribo and xylo-thioadenosine 6 and 10 seems to be just reversed from that of ribo and xyloadenosine as shown by NMR.

Table 1: Characteristic n.m.r. data of the substituted adenosine derivatives ( $\delta$  = ppm, first order coupling constants = Hz at 100 MHz in DMSO-D<sub>6</sub>)

	<u>C<sub>1'</sub>-H</u>	<u>C<sub>2'</sub>-H</u>	<u>Others</u>
<u>4</u>	6.42d J <sub>1',2'</sub> = 2.7	6.18q J <sub>2',3'</sub> = 6	7.74m, 10 protons, phenyl
<u>6</u>	5.98d J <sub>1',2'</sub> = 2.0	4.44q <sup>+</sup> J <sub>2',3'</sub> = 5.0	2.41 br, SH
<u>7</u>	6.26d J <sub>1',2'</sub> = 2.5	5.68m J <sub>2',3'</sub> = 2.5 J <sub>2',3''</sub> = 6	7.74m, 5 protons, phenyl
<u>9</u>	5.92d J <sub>1',2'</sub> = 6.7	4.83q <sup>+</sup> J <sub>2',3'</sub> = 9.2	7.77m, 5 protons, phenyl
<u>10</u>	5.71d J <sub>1',2'</sub> = 6.3	4.55q <sup>+</sup> J <sub>2',3'</sub> = 8.3	2.8 br, SH
<u>12</u>	6.70d J <sub>1',2'</sub> = 9.5	5.58q J <sub>2',3'</sub> = 5.7	6.01q J <sub>3',4'</sub> = 1.2 H <sub>3'</sub> 7.86m, 10 protons, phenyl

<sup>+</sup> = after D<sub>2</sub>O exchange.



When we applied the thiobenzoate displacement reaction to the 2'-deoxy-2'-iodo-arabinofuranosyl adenine derivative 11<sup>7)</sup>, the reaction product again contained a S-benzoyl as well as an O-benzoyl group, as indicated by MS ( $M^+ = 659$  m/e) and n.m.r. (see tab. 1). This product, on catalytic hydrogenation followed by sodium methoxide treatment, gave 2'-deoxyadenosine (13), suggesting structure 12. The  $J_{1,-2}$  coupling constant of 9.5 Hz for compound 12 is in agreement with that assigned by Goodman et al.<sup>12)</sup> for 6-substituted-9-purinyl nucleosides of 2'-thio-D-ribose ( $J_{1,-2} = 9.0 - 9.5$  Hz).

All attempts to remove the protecting groups from 12 with various bases, so far, have only led to isolations of adenine. A similar result has been observed by Goodman et al.<sup>12)</sup> during the deblocking of 6-benzamido-9-(2-S-benzoyl-3,5-di-O-benzoyl-2-thio- $\beta$ -D-ribofuranosyl)-9H-purine.

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#### R E F E R E N C E S

- 1) A. Bloch, "The Design of Biologically Active Nucleosides" in Drug Design, Vol. IV, p. 328, Ed. E.J. Ariens, Academic Press, New York, 1973.
- 2) K.J. Ryan, E.M. Acton, and L. Goodman, J.Org.Chem. 33, 1783 (1968).
- 3) L. Goodman in "Basic Principles of Nucleic Acid Chemistry", Vol. I, p. 134, Ed. P.O.P Ts'0 Academic Press, New York and London, 1974.
- 4) R. Mengel and H. Wiedner, Chem.Ber. 109, 433 (1976).
- 5) R. Mengel and H. Wiedner, Chem.Ber. 109, 1395 (1976).
- 6) M.J. Robins, Y. Fouron, and R. Mengel, J.Org.Chem. 39, 1564 (1974).
- 7) M.J. Robins, R.A. Jones, and R. Mengel, J.Amer.Chem.Soc. 98, 8213 (1976).
- 8) In a similar way coenzyme A was acylated with sodium thioacetate. E.B. Wilson, J.Amer.Chem.Soc. 74, 3205 (1952).
- 9) J.A. McCloskey in "Basic Principles of Nucleic Acid Chemistry", Vol. I, p. 270, Ed. P.O.P. Ts'0 Academic Press, New York and London, 1974.
- 10) The melting point difference might be due to variable disulfide formation.
- 11) G.L. Ellman, Arch.Biochem.Biophys. 82, 70 (1959).
- 12) K.J. Ryan, E.M. Acton, and L. Goodman, J.Org.Chem. 36, 2646 (1971).